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Supporting Information

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# Ligand oxidation of a deprotonated bis(picoly)amine Ir<sup>I</sup>(cod) complex\*\*

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<http://www.chemeurj.org/> or from the author.

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University of Leiden.

**Crystal data for complex K[3]:**  $C_{20}H_{23}IrKN_3$ ,  $M_r = 536.71$ , rough platelet translucent dark red crystal ( $0.21 \times 0.10 \times 0.03$  mm), monoclinic, space group P 21/n,  $a = 12.039(2)$ ,  $b = 10.4751(10)$ ,  $c = 14.844(2)$  Å,  $\beta = 99.197(10)^\circ$ ,  $V = 1847.9(4)$  Å<sup>3</sup>,  $Z = 4$ ,  $\rho_{\text{calcd}} = 1.929$  g cm<sup>-3</sup>,  $F(000) = 1040$ ,  $T = 208(2)$  K, MoK $\alpha$  radiation. Data collected with a Nonius KappaCCD diffractometer. Of 29099 measured reflections ( $2\theta$ : 4.1-50°), 3224 were unique ( $R_{\text{int}} = 0.1139$ ); a multi-scan absorption correction<sup>[s1]</sup> was applied (SADABS program) with min./max. transmission factors of 0.33/0.80. Structure solved by direct methods and difference-Fourier maps; refined using SHELXTL.<sup>[s2]</sup> Final agreement factors were  $R1 = 0.0509$  (2349 obs. reflections,  $F^2 > 2\sigma(F^2)$ ) and  $wR2 = 0.1036$ ; data/restraints/parameters 3224/0/226; GOF = 1.057.

### DFT calculations

The geometry of  $3^-$  was optimized with the Turbomole program<sup>[S3]</sup> coupled to the PQS Baker optimizer.<sup>[S4]</sup> The geometry was fully optimized as a minimum at the at the b3-lyp level<sup>[S5]</sup> using the polarized triple- $\zeta$  def-TZVP basis<sup>[S3c,f]</sup> (small-core pseudopotential<sup>[S3c,e]</sup> on Rh or Ir). Orbitals and spin densities were visualized with the Molden program.<sup>[S6]</sup>

## Selected spectra for the new complexes 2 and K[3].

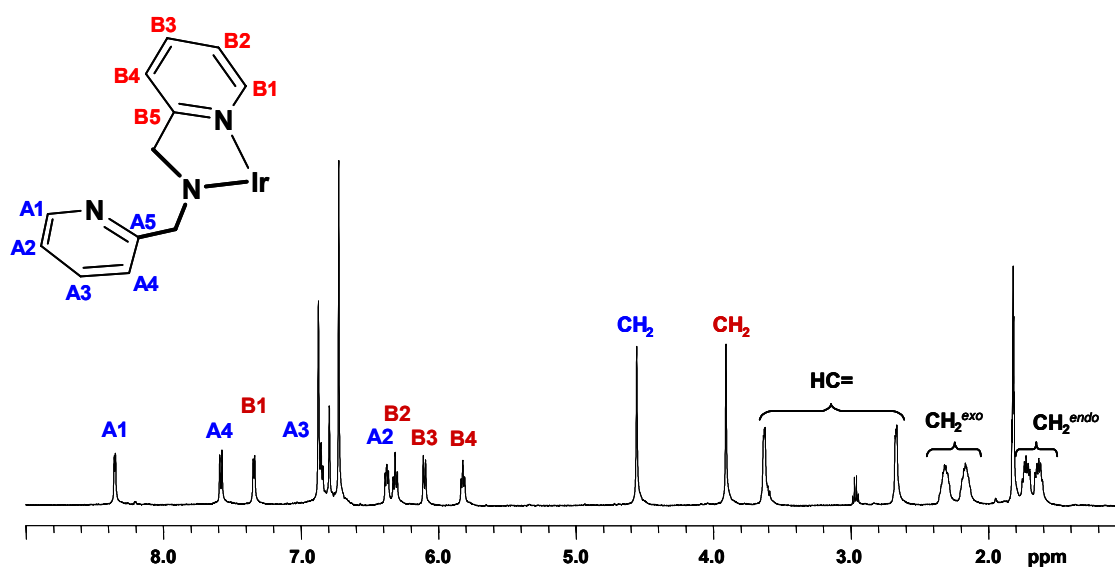


Figure S1.  $^1\text{H}$  NMR ( $d^8$ -toluene,  $-40^\circ\text{C}$ ) of the amido complex  $[\text{Ir}(\text{bpa-H})(\text{cod})]$  (**2**).

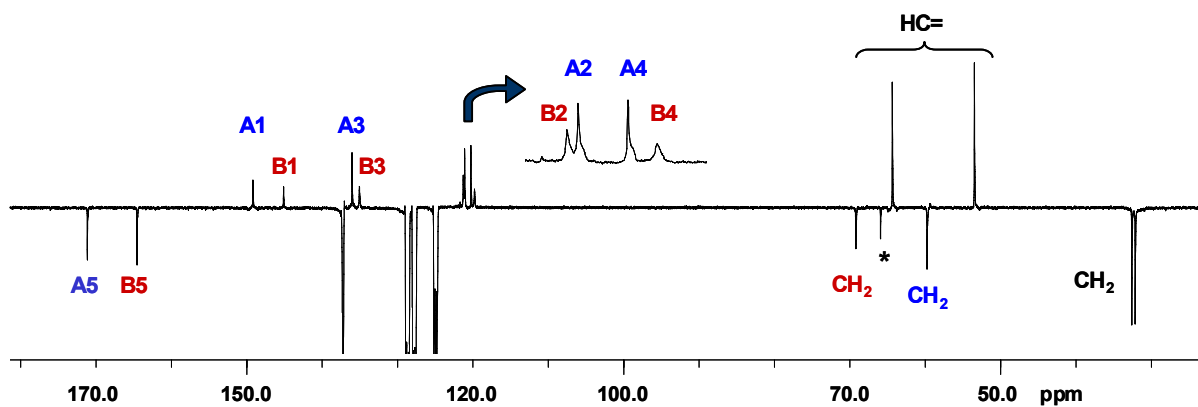


Figure S2.  $^{13}\text{C}\{^1\text{H}\}$ -APT NMR ( $d^8$ -toluene,  $-40^\circ\text{C}$ ) of the amido complex  $[\text{Ir}(\text{bpa-H})(\text{cod})]$  (**2**). \* denotes the  $\text{CH}_2$  of the diethyl ether.

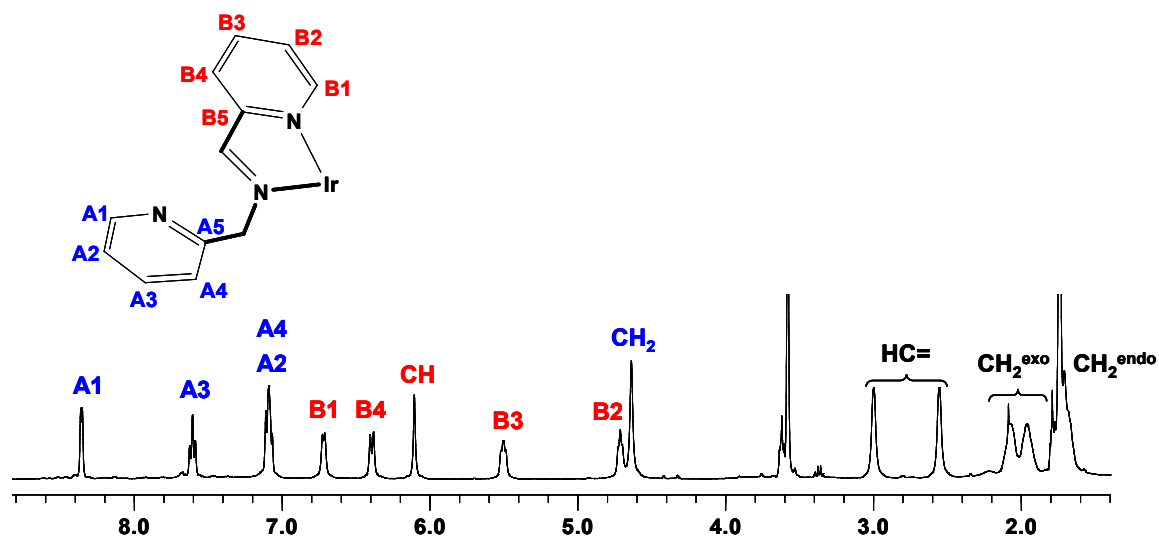


Figure S3.  $^1H$  NMR ( $d^8$ -thf,  $-70$  °C) of the anionic complex  $K[Ir(bpa-2H)(cod)]$  (K[3]).

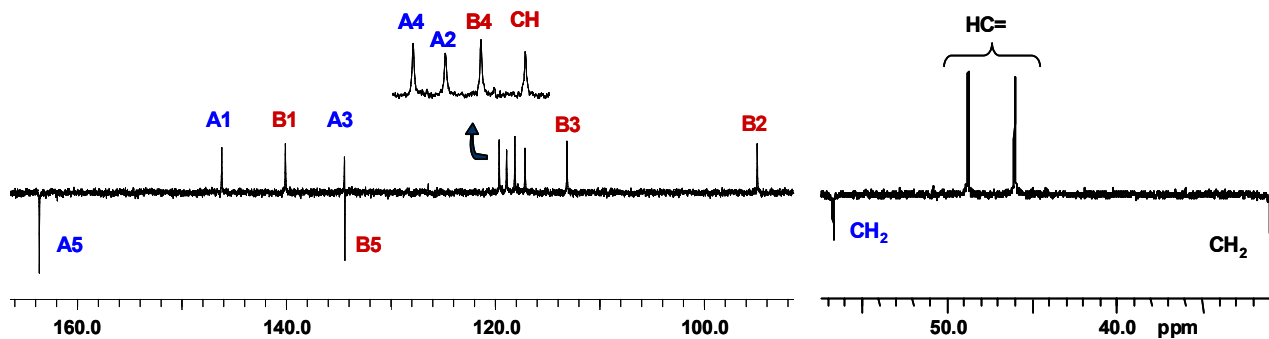


Figure S4.  $^{13}C\{^1H\}$ -APT NMR ( $d^8$ -thf,  $-70$  °C) of the anionic complex  $K[Ir(bpa-2H)(cod)]$  (K[3]).

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